New Compounds: Potential Antidiabetics IV:

1-(2,4-Dinitrophenyl)-3,5-diphenyl-

4-arylazopyrazoles and 1-Carbamoyl-

3,5-diphenyl-4-arylazopyrazoles

## H. G. GARG and PREM PAL SINGH

**Keyphrases** ☐ Antidiabetics—synthesis, pyrazoles, isoxazoles ☐ 1-(2,4-Dinitrophenyl)-3,5-diphenyl-4-arylazopyrazoles—synthesis, potential hypoglycemics ☐ 1-Carbamoyl-3,5-diphenyl-4-arylazopyrazoles—synthesis, potential hypoglycemics

Numerous derivatives of pyrazoles and isoxazoles have been prepared for testing their antidiabetic activity (1–4). The compounds 1-phenyl-3-methyl-4-arylhydrazono-2-pyrazolin-5-ones have shown encouraging antidiabetic activity in experimental animals (5). To examine their hypoglycemic activity, a series of 1-(2,4-dinitrophenyl)-3,5-diphenyl-4-arylazopyrazoles and 1-carbamoyl-3,5-diphenyl-4-arylazopyrazoles have been prepared and are reported in the present communication.

Precursors 1,3-diphenyl-2-arylhydrazono-1,2,3-propanetriones (I) were obtained by coupling diazotized anilines with 1,3-diphenyl-1,3-propanedione (4). 2,4-Dinitrophenylhydrazine and semicarbazide hydrochlo-

Table I—Characteristics of 1-(2,4-Dinitrophenyl)-3,5-diphenyl-4-arylazopyrazoles

		Yield,				Anal., %	
No.	R	%	M.p.	Color	Formula	Calcd.	Found
1	2-Nitrophenyl	55	208-209°	Red crystals	$C_{27}H_{17}N_7O_6$	N, 18.3	N, 18.0
2	3-Nitrophenyl	65	196-197°	Red crystals	$C_{27}H_{17}N_7O_6$	N, 18.3	N, 18.1
3	3-Chlorophenyl	60	203°	Orange	$C_{27}H_{17}ClN_6O_4$	C, 61.8 H, 3.2 N, 15.4	C, 61.6 H, 3.4 N, 15.0
4	4-Chlorophenyl	75	186°	Orange needles	$C_{27}H_{17}ClN_6O_4$	N, 15.4	N, 15.3
5	2-Methylphenyl	70	228229°	Orange needles	$C_{28}H_{20}N_6O_4$	N, 16.6	N, 16.4
6	3-Methylphenyl	55	198-199°	Yellow- orange	$C_{28}H_{20}N_6O_4$	N, 16.6	N, 16.3
7	4-Methylphenyl	50	163°	Red crystals	$C_{28}H_{20}N_6O_4$	C, 66.7 H, 4.0 N, 16.6	C, 66.6 H, 4.1 N, 16.1
8	2-Methoxyphenyl	50	233°	Orange-red needles	$C_{28}H_{20}N_6O_5$	N, 16.1	N, 16.0
9	2-Ethoxyphenyl	60	221-222°	Orange needles	$C_{29}H_{22}N_6O_5$	N, 15.7	N, 15.6
10	4-Ethoxyphenyl	65	132°	Pale- yellow needles	$C_{29}H_{22}N_6O_5$	N, 15.7	N, 15.4
11	4-Sulfamoyl- phenyl	55	132–133°	Orange	$C_{27}H_{19}N_7O_6S$	N, 17.2	N, 17.0
12	2,5-Dichloro- phenyl	75	233°	Orange	$C_{27}H_{16}Cl_2N_6O_4$	Cl, 12.7	Cl, 12.4
13	2,5-Dimethyl- phenyl	60	238-239°	Orange needles	$C_{29}H_{22}N_6O_4\\$	N, 16.2	N, 16.0
14	2,5-Dimethoxy- phenyl	55	242-243°	Dark brown	$C_{29}H_{22}N_6O_6$	N, 15.2	N, 15.0
15	2-Chloro-6- methylphenyl	65	<b>22</b> 4°	Orange	$C_{28}H_{19}ClN_6O_4$	C, 62.4 H, 3.5 N, 15.6	C, 62.2 H, 3.4 N, 15.1

Table II—Characteristics of 1-Carbamoyl-3,5-diphenyl-4-arylazopyrazoles

		Yield,				———Anal., %———	
No.	R	%	M.p.	Color	Formula	Calcd.	Found
1	2-Nitrophenyl	60	193-194°	Orange fibers	C <sub>22</sub> H <sub>16</sub> N <sub>6</sub> O <sub>3</sub>	N, 20.3	N, 20.0
2	3-Nitrophenyl	50	265–266°	Orange	$C_{22}H_{16}N_6O_3$	C, 64.1 H, 3.9	C, 64.1 H, 4.0
3	2-Methylphenyl	50	192–193°	Orange	$C_{23}H_{19}N_5O$	N, 20.3 C, 72.4 H, 4.9 N, 18.3	N, 19.9 C, 72.2 H, 4.8 N, 17.9
4	3-Methylphenyl	45	242-243°	Orange fibers	$C_{23}H_{19}N_5O$	N, 18.3	N, 18.2
5 6	4-Methylphenyl 2-Methoxyphenyl	65 50	219–220° 180–181°	Orange Orange	$C_{23}H_{19}N_5O \\ C_{23}H_{19}N_5O_2$	N, 18.3 C, 69.5 H, 4.8 N, 17.6	N, 18.0 C, 69.4 H, 4.6 N, 17.2
7 8	3-Methoxyphenyl 4-Methoxyphenyl	45 55	157° 168–171°	Red Orange fibers	$\begin{array}{c} C_{23}H_{19}N_5O_2 \\ C_{23}H_{19}N_5O_2 \end{array}$	N, 17.6 C, 69.5 H, 4.8 N, 17,6	N, 17.5 C, 69.6 H, 4.9 N, 17.1
9 10	2-Ethoxyphenyl 4-Sulfamoyl- phenyl	50 60	163-165° 250-251°	Orange Orange	$C_{24}H_{21}N_5O_2 \ C_{22}H_{18}N_6O_3S$	N, 17.0 N, 18.8	N, 17.2 N, 18.5
11	2,5-Dichlorophenyl	70	207-209°	Orange	$C_{22}H_{15}Cl_2N_5O$	Cl, 16.2	Cl, 16.0
12	2,6-Dichloro- phenyl	75	161–162°	Orange	$C_{22}H_{15}Cl_2N_5O$	Cl, 16.2	Cl, 16.3
13	2,5-Dimethyl- phenyl	60	186–187°	Orange needles	$C_{24}H_{21}N_5O$	C, 72.9 H, 5.3 N, 17.7	C, 72.8 H, 5.1 N, 17.2

ride react with I to yield 1-(2,4-dinitrophenyl)-3,5-diphenyl-4-arylazopyrazoles (II) (Table I) and 1-carbamoyl-3,5-diphenyl-4-arylazopyrazoles (III) (Table II) congeners, respectively. They are all highly colored crystalline compounds and are soluble in common organic solvents.

## **EXPERIMENTAL**

Melting points were taken with a Kofler hot stage apparatus and are uncorrected.

**1,3-Diphenyl-2-arylhydrazono-1,2,3-propanetriones** (I)—These were prepared by coupling aryldiazonium salts, prepared from anilines, with 1,3-diphenyl-1,3-propanedione (4).

1-(2,4-Dinitrophenyl)-3,5-diphenyl-4-phenylazopyrazole (II,  $R=C_0H_5)$ —To 1,3-diphenyl-2-phenylhydrazono-1,2,3-propanetrione (0.82 g., 0.0025 mole) in glacial acetic acid (25 ml.) was added 2,4-dinitrophenylhydrazine (0.50 g., 0.0025 mole) in ethyl alcohol (15 ml.) containing concentrated sulfuric acid (2 ml.). The mixture was refluxed for 8 hr. On cooling, shining crystals separated out which were recrystallized from alcohol (0.60 g., 50%) as yellow-orange crystals, m.p. 136–137°.

Anal.—Calcd. for C<sub>27</sub>H<sub>18</sub>N<sub>6</sub>O<sub>4</sub>: N, 17.1. Found: N, 16.8.

The details of the other 1-(2,4-dinitrophenyl)-3,5-diphenyl-4-arylazopyrazoles which were prepared are given in Table I.

1-Carbamoyl-3,5-diphenyl-4-phenylazopyrazole (III,  $R = C_6H_5$ )—A solution of 1,3-diphenyl-2-phenylhydrazono-1,2,3-propanetrione

(0.82 g., 0.0025 mole) in alcohol (30 ml.) was treated with an aqueous solution of semicarbazide hydrochloride (0.23 g., 0.0025 mole). This was heated under reflux for 2 hr. On cooling, shining crystals separated out which were recrystallized from alcohol (0.50 g., 65%) as orange fibers, m.p. 215°.

Anal.—Calcd. for C<sub>22</sub>H<sub>17</sub>N<sub>5</sub>O: N, 19.07. Found: N, 18.8.

The details of the other 1-carbamoyl-3,5-diphenyl-4-arylazopyrazoles which were prepared are given in Table II.

## REFERENCES

- (1) H. G. Garg and P. P. Singh, J. Med. Chem., 11, 1103(1968).
- (2) Ibid., 11, 1104(1968).
- (3) H. G. Garg and P. P. Singh, J. Chem. Soc. (C), 1969, 1141.
- (4) H. G. Garg and P. P. Singh, unpublished data.
- (5) H. G. Garg, D.Sc. thesis, Agra University, Agra, India, 1969.

## ACKNOWLEDGMENTS AND ADDRESSES

Received November 6, 1969, from the Department of Chemistry, University of Roorkee, Roorkee, India.

Accepted for publication January 13, 1970.

The authors thank Dr. Maxwell Gordon, Smith Kline & French Laboratories, Philadelphia, Pa.; Prof. W. U. Malik, Head of Chemistry Department, for providing the necessary facilities for work; and the C.S.I.R., New Delhi for the award of a Junior Research Fellowship (to P.P.S.)